

**Calcium phosphate ceramics and
particles for *in vivo* and *in vitro* transfection**

5 The present invention relates to a method for the
transfection of DNA attached to the surface of calcium
phosphate ceramics having specific characteristics.
This method may comprise a step for preparing a
material in a salt solution or a cell culture medium in
10 order to improve the attachment of DNA and its
availability for the transfection of cells. The
invention also relates to the use of modified calcium
phosphate ceramics and powders for the transfection of
cells *in vitro* and *in vivo* and for the culture of
15 transfected cells in a three-dimensional network.

The transfection of genes into eukaryotic cells is a
key step in gene therapy. Several methods can be used
with variable yields. They can be used *in vitro* or *in*
20 *vivo*.

For the purposes of gene therapy, cells may be
transfected *in vitro* and then reinjected into the body
or directly into the organs or tissues in which they
25 reside (Evans, C.H., Robbins, P.D., Possible
orthopaedic applications of gene therapy, J Bone Joint
Surg, 77-A, 7:1103-1114).

The various methods used for cell transfection are
30 summarized in the table below:

Method	Advantages	Disadvantages
DEAE-dextran	Simple	Transient expression
Calcium phosphate	Simple	Unusable for cells in suspension
Liposomes	Simple	Relatively unproven
Microinjection	Efficient	Technically difficult
Electroporation	Good for nonadherent	No cotransfection

	cells	
Protoplast fusion	Good for nonadherent cells	Variable results
Adenoviruses	High infectivity, unknown production, infects nondividing cells, large variety of host cells	DNA integrated as episome, toxic, production of viral proteins
Adeno-associated viruses	Nonpathogenic, stable expression, infects nondividing cells, large variety of host cells	Supports only short genes, difficult to produce, not highly developed
Herpes simplex	Infects nondividing cells, supports long genes	Toxic, transient expression, not highly developed
Retrovirus infection	Efficient	Cell type reduced by tropism, low coding capacity
Polycationic solids	Simple, localized transfection	Transient expression
Satellite chromosome	Makes it possible to transfect long genes	Results unproven
Others: polymers in the form of a hydrogel, polycationic lipids, polylysine, polyornithine, histones and other chromosomal proteins, hydrogenated polymers		Low yields, variable results, uses in vivo difficult, variable biocompatibility

Approximately fifteen years after clinical trials of gene therapy started, the results have been
5 disappointing overall for several reasons:

Regardless of the vectors used, adenoviruses, adeno-associated viruses (AAV), retroviruses or physicochemical formulations, the efficiency of gene
10 transfer into the target cells has always been very low [A. Kahn. Dix ans de thérapie génique: deceptions et espoirs (Ten years of gene therapy: disappointments and hopes). *Biofutur* 202:16-21, 2000]. The duration of expression of the therapeutic transgenes is most often
15 brief, limited to a few weeks, because of an immune reaction which causes the preferential elimination of the transduced cells, the intrinsic longevity thereof or the extinction of DNA sequences or promoters which direct the expression of the inserted genes (Orkin,
20 S.H., Motulsky, A.G., report and recommendations of the panel to assess the NIH investment in research gene therapy.www.nih.gov/news/panelrep.html).

Finally, certain vectors have manifested a toxic
25 effect. Accidents have occurred during the use of adenoviral vectors injected into the body which caused the death of patients in trials for treatment with ornithine transcarbamylase (Smaglik, P., Investigators ponders what went wrong after gene therapy death. The
30 Scientist 13 [21] : 1 (1999).

Thus, it is evident from the analysis of all the clinical trials of gene therapy that the strategy of transfer of a gene would require vectors which are a lot more efficient, safer and capable of preferentially
35 transfecting the cells on which a therapeutic effect is necessary (Orkin, S.H., Motulsky, A.G., report and recommendations of the panel to assess the NIH investment in research gene therapy.

www.nih.gov/news/panelrep.html).

It is for this reason that polycationic polymeric vectors were developed. These vectors are solids and
5 can adsorb DNA in various forms, in particular in the form of a plasmid. They have the characteristic feature of transfecting cells coming into contact with them with a variable yield. They have been used *in vivo* to transfect cells of the loose connective tissues
10 involved in bone wound healing in order to accelerate it (S. Goldstein and J. Bonadio. *in vivo* gene transfer methods for wound healing. The Regent of the University of Michigan. Anonymous. United States:(5,962,427):1-31, 1999. Gene Therapy. A61K 48/00. 514/44).

15 Coprecipitates of calcium phosphates and DNA have been used for many years to transfect cells *in vitro* (E.T. Schenborn and V. Goiffon. Calcium phosphate transfection of mammalian cultured cells. edited by
20 M.J. Tymms, Totowa, NJ:Humana Press Inc, 2000, p. 135-144; W. Song and D.K. Lahiri. Efficient transfection of DNA by mixing cells in suspension with calcium phosphate. *Nucleic Acid Research* 23 (17):3609-3611, 1995; Y.-W. Yang and J.-C. Yang. Calcium phosphate as a
25 gene carrier: electron microscopy. *Biomaterials* 18:213-217, 1997).

They are obtained by pouring a calcium chloride solution into the medium in order to supersaturate it
30 with calcium and to precipitate a calcium phosphate in which DNA molecules are embedded. These composite particles are then phagocytosed by the cells which integrate the plasmid in various ways and express the genes which are transported.

35 However, these coprecipitates have a major disadvantage. They are very difficult to use *in vivo* because it is difficult to obtain supersaturation in an

open system. Moreover, they cannot allow localized transfections in space.

5 Calcium phosphate ceramics are materials obtained by sintering a slip containing particles of calcium phosphate in suspension. They are assemblages of grains linked by grain boundaries (Frayssinet, P., Fages, J., Bonel, G., Rouquet, N., Biotechnology, material sciences and bone repair. European Journal of Orthopaedic
10 Surgery & Traumatology (1998) 8:17-25).

These materials exhibit a specific biocompatibility with the bone tissue, which makes them particularly useful as bone reconstruction materials or as vector
15 for osteogenic cells (P., Frayssinet, J.L. Trouillet, N. Rouquet, E. Azimus, A. Autefage (1993), Osseointegration of macroporous calcium phosphate ceramics having a different chemical composition. Biomaterials, 14, 6: 423-429).

20 In the context of the invention, we have developed calcium phosphate ceramics and powders capable of transfecting cells both *in vivo* and *in vitro*, in particular mesenchymatous cells. The chemical
25 composition of these ceramics may vary because several orthophosphoric acid salts can enter into their composition, in particular tricalcium phosphate, hydroxyapatite which is the synthetic phase closest to the mineral phase of the bone tissue, and octocalcium
30 phosphate. These ceramics have another characteristic feature, they have surface properties which are highly variable according to various parameters such as, *inter alia*, the mode of synthesis of the powder, the baking temperature or the presence of various trace elements.
35 These various factors influence in particular the surface charge, the zeta potential and the capacities for substitution in the mesh of the calcium phosphate. Calcium phosphate ceramics also have the characteristic

feature of exhibiting epitaxial carbonated apatite growth at their surface once implanted in the body or immersed in a salt medium of comparable composition to the extracellular fluid (M. Heughebaert, R.Z. LeGeros, 5 M. Gineste and A. Guilhem. Hydroxyapatite (HA) ceramics implanted in non-bone-forming sites. Physico-chemical characterization. *J Biomed Mat Res* 22:257-268, 1988). It is to these crystalline growths that the biocompatibility properties of these materials have 10 been attributed.

The adsorption properties of calcium phosphates in relation to nucleic acids have been exploited in chromatography on HA columns to separate and purify DNA 15 or certain RNAs. It is important to understand that given the same chemical composition, all the hydroxyapatite powders used in chromatography do not have the same nucleic acid separating power (A. Eon-Duval, Purification of plasmid DNA by hydroxyapatite 20 chromatography, Abstract of 2nd conference on hydroxyapatite. San Francisco March 2001). The interactions between organic molecules and hydroxyapatite depends on the surface properties of this mineral (M.J. Gorbunoff, Protein chromatography on hydroxyapatite columns. 25 *Methods in Enzymology*, vol 182, Academic Press Inc 1985: 329-339), which can vary from one batch to another.

It has been proved that the distribution of the charges 30 at the surface of the solid and its hydration capacities have a great influence on the adsorption of organic molecules at its surface (Norde, W., Lyklema, J., (1991) Why proteins prefer interfaces. *J Biomed Sci Polymer Edn* 2, 183-202 (1991)). Likewise, the ionic 35 strength and the pH of the solvent for organic molecules should be taken into account.

If the protein in solution and the solid have opposite

charges, they attract each other. At least if the charge on the protein and that on the surface of the solid roughly balance. If the charges do not balance, that results in an accumulation of charges in the
5 region of contact, causing a high electrostatic potential which is energetically not very favorable for an adsorption. A similar situation is observed when the surface of the solid and the organic molecule are of the same sign. However, in numerous cases, the
10 adsorption may occur nevertheless in some cases by virtue of the incorporation of ions from the solution at the interface of the adsorbed layer which prevents the accumulation of charge.

15 The hydrophobicity has an influence on the adsorption because it participates in the distribution of the charges, in particular in organic molecules which have a tertiary and quaternary structure. The hydrophobicity of a surface (molecule or solid) can promote
20 adsorption.

The distribution of the charges and the hydration capacities of apatites are advantageous properties because they may have a positive or negative surface
25 charge and may be hydrophilic or hydrophobic. Furthermore, as substitutions in the mesh may be numerous, the functional groups at the surface can vary.

30 We have developed hydroxyapatite-based calcium phosphate powders capable of binding DNA in various forms and of delivering it to isolated cells or into the body for transfection purposes. These powders may be injected in suspension in a liquid or a gel. They
35 may also be deposited using a spoon or serve as a transfecting vector for cells cultured in a three-dimensional network. They have specific physico-chemical properties in order to possess these trans-

fection properties. A series of experiments were carried out which make it possible to judge the transfection of isolated or nonisolated cells with a plasmid carrying the gene for galactosidase which can be demonstrated by histochemistry. The powder is an embodiment which is particularly well suited to be able to transfect both isolated cells and tissues both *in vitro* and *in vivo*. These powders allow internalization of the DNA and its protection from intracytoplasmic nucleases and its transfer into the nucleus.

The mechanism involved in the attachment of DNA (an organic molecule having a negative charge) to the surface of hydroxyapatite particles may be:

- An electrostatic adsorption when the material is positively charged.
- A coprecipitation of the DNA molecules in the carbonated apatite layer appearing through epitaxial growth at the surface of these materials and resulting from complex dissolution/reprecipitation processes occurring at the surface in media supersaturated with calcium and phosphorus.
- An ion exchange between the interfacial phase and the solution.

Once attached to the material, the DNA should penetrate into the cell. The composition and the surface characteristics are also important for the degradation of the material in a biological medium and the emission of transfecting particles. It is known that HA ceramics undergo degradation at the grain boundaries and that the carbonated apatite layer appearing at the surface of the material by epitaxial growth has a different solubility from the material itself.

On the other hand, all calcium phosphates cannot transfect cells. DCPD for example or certain embodiments of

HA or TCP have shown their incapacity to do it. Their cytotoxicity is certainly responsible for that.

5 By contrast, the modification of the surface of the powders and the ceramics by maturation in a culture medium causing epitaxial carbonated apatite growth improves the labeling yield.

Description

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Thus, in a first aspect, the present invention relates to a method for creating a mineral-DNA composite, characterized in that it comprises a step consisting of an incubation in a salt or culture medium not saturated with calcium and phosphorus in the presence of the DNA molecule. This method makes it possible to obtain attachment of DNA to the surface of the ceramic by adsorption onto a ceramic surface modified by epitaxial growth or by coprecipitation at the surface of the material. These calcium phosphate particles are immersed in a salt medium or a culture medium of the cell culture media type commonly used in biotechnology, in particular DMEM, for a few minutes, for example 1, 5, 10 or 30 minutes at least to about 12, 24, 48 hours, a few days or more at a temperature ranging from 15 to 50°C, preferably about 37°C. The aim is to have the formation of a carbonated apatite layer at the surface before or during the bringing into contact with the plasmids.

30

In a particular embodiment, the method mentioned above is carried out before the bringing into contact with the nucleic acids, in particular plasmids. Alternatively, this step which causes epitaxial carbonated apatite growth at the surface of said powders and ceramics is carried out in a medium containing the nucleic acids. In this mode, the surface modification and the attachment of the nucleic acids

are performed simultaneously.

Preferably, the powders and ceramics are immersed in a DMEM culture medium for 48 hours at 37°C before or simultaneously with the attachment of the nucleic acids.

In an additional aspect, the invention relates to a method for attaching DNA in plasmid form to the surface of calcium phosphate ceramic or powder, characterized in that it comprises a step a) consisting of a hydration of the calcium phosphate powder or calcium phosphate ceramic in a phosphate buffer solution not saturated with calcium and phosphate and a step b) consisting of an immersion of the products obtained in step a) in a phosphate buffer solution not saturated with calcium and phosphate containing a single- or double-stranded DNA for periods varying from a few minutes to several hours, c) producing calcium phosphate particles containing DNA molecules attached to its surface.

Preferably, the solution in step a) and b) comprises a 0.12 M phosphate buffer (pH 6.8). The immersion is carried out for at least 1, 5, 10 or 30 minutes up to about 12, 24 or 48 hours at a temperature ranging from 15 to 50°C, preferably about 37°C. In addition, the calcium phosphate particles are kept immersed in a culture medium of the cell culture media type, for about a few minutes to a few days, and at a temperature ranging from 15 to 50°C, preferably about 37°C.

Thus, in this method, the hydration preferably consists of an immersion of the calcium phosphate powder or of the calcium phosphate ceramic in a solution simulating the extracellular fluids intended to produce an epitaxial carbonated apatite growth at the surface of said powders and ceramics. In this regard, step b) is carried out by means of a medium simulating the extra-

cellular fluids or a medium of the cell culture media type containing the nucleic acids, said medium being nondenaturing for the DNA and not saturated with calcium and phosphate. This medium causes epitaxial
5 carbonated apatite growth at the surface of said powders and ceramics.

Steps a) and b) may be carried out simultaneously or successively. Thus, it is possible to carry out the
10 invention with a solution containing a single- or double-stranded DNA for periods varying from a few minutes to several hours at about 37°C.

Advantageously, this method makes it possible to attach
15 the DNA at physiological pH to calcium phosphate particles under conditions which are not denaturing for the DNA molecule. The ceramics may be porous or dense ceramics.

20 In another aspect, the invention relates to a method for transfecting isolated cells, cultured in a monolayer or in three dimensions, consisting of bringing the cells to be transfected into contact with the particles obtained by the method described above
25 for periods of a few hours to a few weeks. This method can also be carried out to transfect cells contained in a cultured tissue fragment. The particles obtained which are mentioned above are particularly useful for the preparation of a medicament for transfecting in
30 vivo cells contained in a tissue or in an organ.

In another aspect, the invention relates to the calcium phosphate ceramics and powders which can be obtained from the method described above, characterized in that
35 they can support epitaxial carbonated apatite growth at their surface under nondenaturing conditions, in particular in a salt solution which is not saturated and nondenaturing for the biological macromolecules.

The invention also relates to these calcium phosphate ceramics and powders additionally comprising the nucleic acids attached to their surface.

5

These products are particularly efficient for the transfection of cells *in vitro* and *in vivo*.

Advantageously, the powders and ceramics obtained
10 possess at least one of the properties described below before the surface modification:

- nature of the charged groups at the surface: PO_4^- , OH^- , Ca^{++}
- basic surface pH
- 15 - negative electrokinetic potential
- hydrophobic
- particle size between 0-200 μm , in particular between 80-125 μm and 0-25 μm .

20 Preferably, the products of the invention comprise all the characteristics described above.

In addition, the calcium phosphate ceramics and powders mentioned above may contain a core composed of another
25 polymeric, ceramic or metallic, preferably magnetic, material.

The invention also relates to the formed particles based on calcium phosphate powders described above,
30 said particles being contained in a mineral or polymeric matrix, in particular in calcium sulfate or phosphate cements.

In another aspect, the invention relates to a ceramic
35 coating for joint prostheses having the characteristics of the ceramic defined above.

The invention also relates to the use of said calcium

phosphate ceramics and powders loaded with DNA at their surface as a support for cell culture, in particular for the culture in a three-dimensional network of cells transfected with the support and for the transfection of cells *in vitro* and *in vivo*.

The following examples are given by way of illustration. They constitute preferred embodiments of the invention.

Example 1: Characteristic of the powders used

Type P15: spherical powder having a specific surface area of $0.62 \text{ mg}^2/\text{g}$. They were calcined at 1180°C and their particle size is between $80\text{-}125 \text{ }\mu\text{m}$.

Type P1: powder of any shape having a specific surface area of $56.84 \text{ m}^2/\text{g}$, not calcined (crude), having a particle size of between $0\text{-}25 \text{ }\mu\text{m}$.

The study of the particle size of the powders used shows that the spherical powders (P15) have a well defined particle size range whereas those having any shape (P1) have much broader particle size ranges with a lot of fine particles. The zero charge pH varies with the powder calcination temperature. The zeta potential of the powder P1, measured in demineralized water, is -27.5 mV and the surface pH is 9.08.

Depending on the powder sintering temperature, the zero charge pH is variable but considerably less than the physiological pH. This means that regardless of the sintering temperature, the electrokinetic potential of the powders, at neutral pH, is negative.

Examination of the spherical powders by scanning

microscopy shows that they consist of grains assembled through grain boundaries. Surface irregularities exist on some of the faces of the grains at high magnification.

Powder	P1	P15
Nature of the charged groups	PO_4^- , OH^- , Ca^{++}	PO_4^- , OH^- , Ca^{++}
Electrokinetic potential (mV)	-27.5	-35
Hydrophobicity	+	+
Surface pH	9.8	7.8
Particle size (μm)	0-25	80-125
Specific surface area (m^2/g)	56.84	0.62
Zero charge pH		
Shape of the powder	angular	spherical

Example 2: Method of attaching DNA to the vector

5 The vector may be used in two different ways:

Method A: It may be incubated directly with the plasmid in a phosphate buffer solution. It is then kept incubated therein for several hours while its surface is modified by epitaxial carbonated apatite growth. The attachment may then occur by coprecipitation at the surface of the material.

Method B: It may also be exposed to a salt solution for several days in order to modify the surface. Once the latter has reached equilibrium, the material is then placed in the solution containing the plasmid. The attachment of the DNA is assumed to then occur at the surface of the modified material.

Attachment of the plasmid to the surface of the native particles (method A):

Double-stranded DNA has a marked affinity for HA when it is dissolved in low concentrations of phosphate buffer. They are eluted in higher concentrations of phosphate buffer. 1 m^2 of surface area of powder was deposited in Petri dishes, that is 1.61 g for type A and 0.017 g in type B.

- Hydration of the HA powder (2 ml/g) in 10 ml of 0.12 M phosphate buffer at pH 6.8. Heating for 15 to 30 min at 100°C.

5 • Allow to stand at room temperature and remove the buffer. Resuspend in 5 to 10 ml of 0.12 M phosphate buffer at pH 6.8 at 60°C, separate by decantation and resuspend in 5 ml of the same buffer at 60°C.

10 • Add the nucleic acid sample in 1 ml of 0.12 M phosphate buffer at pH 6.8 at 40°C (the elution of the double-stranded nucleic acids can be carried out by washing the HA 8 to 10 times with 0.5 ml of phosphate buffer (0.4 M)).

15 **Attachment of the plasmid to the surface of the particles modified by epitaxial growth (method B):**

- The particles were incubated at 37°C in DMEM culture medium for 48 hours.

20 • They are washed in a 0.12 M phosphate buffer solution at pH 6.8

- The nucleic acid sample in 1 ml of 0.12 M phosphate buffer at pH 6.8 is added at 40°C.

25 **Example 3: Transfection of cells in vitro**

Three lines were used:

- Rabbit growth cartilage
- Rabbit periosteum
- 30 • Rat calvaria cells

They are obtained by digesting the collagenic matrix in a collagenase solution followed by centrifugation.

3.1 Material with unmodified surface

35 The quantity of powder (type B) was still the same: 10 mg.

During the transfection, the cells were not confluent. The cells were transfected at D0 and the first

evaluation by histochemistry of the expression of galactosidase was made at D4, D21, D30.

At D4:

5 All the lines have labeling zones. In the wells trans-
fected with particles, the labeled cells are grouped
around particles although some were nevertheless
distant from them. This distance can be explained by
the fact that the particles emit debris with a high
specific surface area. They can be seen under the
10 microscope in the middle of groups of labeled cells.
Growth cartilage cells: in absolute value, it is the
series which was labeled the most.

At D21

15 As regards the cartilage cells, the number of trans-
fected cells is high. The rat calvaria cells are highly
positive.

At D30

20 The cells have a relative contact inhibition, they are
practically in three dimensions and rounded. Most of
the cells of the three groups are positive. The number
of positive cells and the preceding growth rate appear
to indicate that the plasmids are transmitted from one
cell to another or alternatively that the release of
DNA particles extends over time, the percentage of
25 positive cells would have been very low in the opposite
case. It is also possible that the release of trans-
fecting particles is gradual. The cells preferentially
labeled are those in contact with the particles.

30 Percentage of labeled cells
as a function of the lines used:

Time (days)	calvaria	Conjugation cartilage	periosteum
4	15	32	27
21	39	42	35
30	65	71	60

3.2 Material with a surface modified by epitaxial growth:

From the early periods of culture, most of the cells are labeled.

3.2.1. Transfection on either side of a semipermeable membrane

The grains were deposited in contact with the cells or separated therefrom by a porous membrane (0.2 μm) made of polycarbonate separating them from the cellular lawn. Cellular labeling with galactosidase is evaluated by histochemistry on D4.

The cells in direct contact with the particles are labeled sporadically. The cells which are not in contact with the particles (separated by the membrane) are also labeled. Transfecting particles less than 0.2 μm in size therefore exist which pass across the pores of the polycarbonate membrane.

3.2.2 Transfection of cells in a three-dimensional network

The cell lines described above are suspended in the culture medium. The bed is placed at the bottom of a culture dish. The suspension serves to inoculate a bed of microbeads ($1.5-2 \times 10^5$ cells/0.05 g of beads) which are vectors for plasmids carrying the galactosidase gene. The bed is placed at the bottom of a culture dish. The cells are cultured for 10 to 15 days. The formation of a three-dimensional cellular layer bridging and agglomerating the beads is obtained. This layer also contains an abundant collagenic matrix.

On the observation date, the cells form a three-dimensional network bridging the various particles and assembling them. Optical microscopy reveals that the

cells contained in the cluster of particles are labeled with galactosidase.

5 **3.2.3 Transfection of cells in tissue cultures**

Material used for the labeling: Type A: spherical powder having a specific surface area of $0.62 \text{ m}^2/\text{g}$. They were calcined at 1180°C and their particle size is 80-125 μm . The quantity of powder is a few tens of
10 particles per dish (P15).

A few beads were placed in contact with the bone fragments after having been incubated without pre-immersion (method A).

15

The bone fragments are obtained from femurs, tibias and calvaria of 3-day-old newborn rats. The bone pieces were cleaned to remove the accompanying soft tissues. The long bones were cut into three pieces: 2 epiphyses
20 and the diaphysis. The calvarias were cut into small fragments with sides of 2 to 3 mm. These various fragments were deposited at the surface of a 3% agar gel in DMEM. The culture medium (DMEM+FCS) was then added so that the fragments show at the liquid-air
25 interface.

The beads were kept in contact with the tissues for 2 to 30 days, on which date the galactosidase activity of the cells was demonstrated before preparing
30 histological sections.

At 2 days of contact, sporadic labeling zones are identifiable. The labeling is made at a distance and in contact with HA beads. It also takes place in contact
35 with these same beads.

At 30 days, the entire bone fragments changed to blue macroscopically (**figure 1**).

Figure 1 represents a macrophotograph of a bone tissue culture in the presence of transfecting powder for 30 days. The bone fragment is completely blue because of the transfection of the cells with the galactosidase vector plasmid. Using reflection optical microscopy, it is not possible to see a zone which is not labeled. The beads are stuck in a matrix labeled by the reaction with galactosidase.

10 The sections of the various bone tissue samples cultured for 30d show that the bone cells (osteoblasts, chondroblasts, perichondral cells, periosteal cells, osteoclasts) are labeled (**figure 2** is a histological section of the same tissue showing that all the cells were transfected with galactosidase X 30).
15 The hematopoietic cell lines are not labeled. It should be noted that:

- all the bone cells are labeled
- they are labeled regardless of the distance from
20 the cells to the beads.

3.2.4 Transfection in vivo

A group of ten 4-week-old male NZW rabbits is selected randomly. These rabbits are divided into two groups:
25 group A and group B. One rabbit from each group serves as control.

The operating zone is located on the left mandible behind the mandibular incisors. It should be noted that a preliminary study made it possible to select this
30 site in which the bone is the most abundant. The P15 powder type was used. The DNA was attached by method A.

After the fitting of sterile fields and skin and mucosal disinfection, an intrabuccal vestibular
35 incision is made using a bistoury. A strip of full thickness is inclined in order to reach the mandibular bony zone at the base of the incisors. A 3 mm trephine is used to systematize the bone diffraction. The bone

defect made is of the order of 2 mm in depth. The bone flap is removed with the aid of a bone scissors. The biomaterial is sucked up with the aid of a 5 ml syringe and deposited in the bone defect such that it fills it.

- 5 A slight pressure is used with a sterile gauze in order to keep the biomaterial in place. The repositioned flap is then sutured.

Two controls undergo a second contralateral operation without deposition of biomaterial.

- 10 The rabbits were sacrificed at 3 to 6 weeks. The mandibles were removed, fixed in ethanol and embedded in hydroxyethyl methacrylate. Sections 5 μ m thick were made and the galactosidase activity demonstrated.

- 15 • At 3 weeks:

In the control sites:

- The histological sections show a spongy bone with few trabeculae whose pores are occupied by a very loose stromal tissue. Multinucleated cells of osteoclastic appearance exist at the surface of the trabeculae. These cells are all labeled by the galactosidase reaction. In the same manner, all the monocytes are also labeled. They are the only cells which are labeled.
- 20

- 25 *In the implanted sites:*

- The sections passing through the calcium phosphate beads show that the beads are embedded in a relatively dense connective tissue with numerous multinucleated cells at their surface. All the fibroblastic or multinucleated cells are labeled with galactosidase.
- 30

- When the sections become distant from the beads, fewer labeled cells exist; nevertheless, the tissue structures which were not disrupted by the surgical procedure give interesting information. The fibroblasts of the dental ligaments are labeled. Cell islets with a fibroblastic appearance exist which are labeled in the stromal tissue of the pores between the trabeculae. In some cases, it even appears that cells of the osteo-
- 35

blastic line are also labeled.

• 6 weeks:

Macroscopically labeling exists around the HA grains.

- 5 The sections show positive stromal cells, the cells of the dental ligaments and the odontoblasts express the gene.